

Restrictive Flow Orifices

The Restrictive Flow Orifice (RFO) is for use in conjunction with high purity compressed gas applications in both the semiconductor and allied chemical industries. Matheson Tri-Gas pioneered the development of RFO technology.

Restrictive Flow Orifices are used to limit the potential danger of an uncontrolled flow from a compressed gas cylinder. Unchecked, the instantaneous flow from a 44 liter compressed gas cylinder filled to 2,000 psig can be as much as 20,000 liters per minute. By inserting an RFO into the outlet of the CGA connection the flow rate could be reduced by a factor of 100 to approximately 200 liters per minute.

The Restrictive Flow Orifice is designed to thread into the outlet of most CGA connections that have external male threads. This would include the family of DISS face seal connections (CGA 630 and 710) that are used in high purity semiconductor applications.

The RFO has no moving parts. It is about 3/8" long and is generally constructed of 316LSS. The orifice opening usually varies from 0.006" to 0.060". It is possible to have orifices that are as small as 0.004" and as large as 0.150" depending upon the application.

The orifice is generally unfiltered. A KEL-F gasket is provided as part of the assembly to help create a seal between the restrictor and the valve body. Refer to Figure 1.

The flow rate through an orifice is a function of the following variables:

- Pressure
- Temperature
- Specific Gravity
- Orifice Opening

Correlations assist in predicting the flow of a particular gas or mixture through an RFO. This is done by first determining the flow through the same RFO at the required pressure with a reference gas and then adjusting the specific gravity accordingly.

The pertinent equation is presented below.

$$\text{Flow} = \frac{\text{The Flow Rate of N}_2 \text{ at the Same Pressure}}{\sqrt{\text{Specific Gravity}}}$$

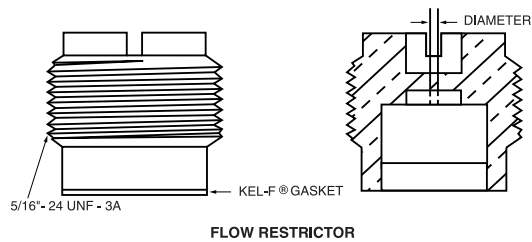


Figure 1: Restrictive Flow Orifice

Typical nitrogen flows for different orifice signs at different pressures can be found in Table I.

By using this table, a reference flow rate for nitrogen can be determined and then the reference equation can be used with this value.

For a mixture a "weighted" or "averaged" specific gravity can be used by multiplying the volume or mole fraction of each component by its specific gravity and then summing.

Table II summarizes some of the gases that can be offered with restrictive flow orifices.

In addition to providing safety features that have been mandated by various local and state codes, the restrictive flow orifice also results in a lower flow rate, which is desirable in many semiconductor applications. While the RFO was not designed to be a modulator or control valve, it will as its name implies, substantially reduce the flow rate under a specific set of pressure and temperature conditions for a given gas.

More detailed information on restrictive flow orifices is available from your local Matheson Tri-Gas Sales Engineer.

TABLE I Orifice Flow Rate at Varying Pressures

| Orifice Size | Pressure-PSIG | | | |
|--------------|---------------|----------|----------|---------|
| | 0 | 500 | 1000 | 2000 |
| 0.006 in. | 0 slpm | 7 slpm | 15 slpm | 29 slpm |
| 0.010 in. | 0 slpm | 18 slpm | 36 slpm | 73 slpm |
| 0.031 in. | 0 slpm | 150 slpm | 310 slpm | — |
| 0.040 in. | 0 slpm | 245 slpm | 485 slpm | — |
| 0.052 in. | 0 slpm | 535 slpm | — | — |

TABLE II Gases offered with Restrictive Flow Orifices Specific Gravity @ 70° (1 atm), CGA and DISS Connections

| Gas | Specific Gravity | CGA | | Gas | Specific Gravity | CGA | | Gas | Specific Gravity | CGA | |
|-----------------------|------------------|-----|------|-------------------|------------------|-----|------|--------------------------|------------------|-----|------|
| | | STD | DISS | | | STD | DISS | | | STD | DISS |
| Ammonia | 0.593 | 660 | 720 | Halocarbon 116 | 4.820 | 660 | 716 | Nitrogen Trifluoride | 2.460 | 670 | 640 |
| Argon | 1.376 | - | 718 | Halocarbon 12 | 4.262 | 660 | 716 | Nitrous Oxide | 1.528 | 326 | - |
| Arsenic Pentafluoride | 6.090 | - | 642 | Halocarbon 13 | 3.610 | 660 | 716 | Oxygen | 1.105 | 540 | 714 |
| Arsine | 2.718 | - | 632 | Halocarbon 14 | 3.075 | - | 716 | Perfluoropropane | 6.652 | 660 | 716 |
| Boron Trichloride | 4.045 | 660 | 634 | Halocarbon 23 | 2.436 | 660 | 716 | Phosphine | 1.190 | 350 | 632 |
| Boron Trifluoride | 2.375 | 330 | 642 | Helium | 0.138 | - | 718 | Phosphorus Pentafluoride | 4.310 | 330 | 642 |
| Carbon Dioxide | 1.527 | 320 | 716 | Hydrogen | 0.070 | 350 | 724 | Silane | 1.120 | 350 | 632 |
| Carbon Monoxide | 0.967 | 350 | 724 | Hydrogen Bromide | 2.780 | 330 | 634 | Silicon Tetrachloride | 5.833 | - | 636 |
| Chlorine | 2.479 | 660 | 728 | Hydrogen Chloride | 1.266 | 330 | 634 | Silicon Tetrafluoride | 3.615 | 330 | 642 |
| Diborane | 0.950 | 350 | 632 | Hydrogen Fluoride | 0.689 | 670 | 638 | Sulfur Hexafluoride | 5.105 | - | 716 |
| Dichlorosilane | 3.473 | 678 | 638 | Hydrogen Selenide | 2.771 | 350 | 632 | Trichlorosilane | 4.666 | - | 636 |
| Disilane | 2.149 | - | 632 | Hydrogen Sulfide | 1.192 | 330 | 722 | Tungsten Hexafluoride | 10.674 | 670 | 638 |
| Germane | 2.634 | 350 | 632 | Krypton | 2.900 | - | 718 | Xenon | 4.558 | - | 716 |
| Halocarbon 11 | 4.770 | 660 | - | Neon | 0.696 | - | 718 | | | | |
| Halocarbon 115 | 5.568 | 660 | 716 | Nitrogen | 0.967 | - | 718 | | | | |